

My First Science Kit

Make a fun crazy dough
and shiny messages!



Science4you



WARNING:

CHOKING HAZARD - Toy contains small parts. Not for children under 3 years.

Play MonSteR



Dear Parents and Guardians:

Through play, children develop different cognitive skills. Scientific studies show that when we are having fun or making discoveries during an experiment, a neurotransmitter called Dopamine is released.

Dopamine is known to be responsible for feelings like motivation, reward and learning and that's why experiences are related to positive feelings. So, if learning is a positive experience, it will stimulate the brain to develop various skills.

Therefore, Science4you aims to develop educational toys that combine fun with education by fostering curiosity and experimentation.

Find out below which skills can be developed with the help of this educational toy!



The educational feature is one of the key strengths of our toys. We aim to provide toys which enable children's development of physical, emotional and social skills.

Find out more about the Brain Activator in Science4you toys at:

www.playmonster.com

PLAYMONSTER



We wanna hear how much fun you had! Get in touch at:

Customer Service

1400 E. Inman Pkwy., Beloit, WI 53511

playmonster@playmonster.com | 1-800-524-4263

For more fun, visit playmonster.com

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Location: United Kingdom

Author: Flávia Leitão

Co-author: Nereida Ribeiro

Scientific Reviewer: Flávia Leitão

Reviewer: Nereida Ribeiro and Flávia Leitão

Conformity reviewer: Luísa Chocahero

Project management: Flávia Leitão

Product development: Flávia Leitão and Ana Garcia

Design management: Marcos Rabelo

Packaging design: Filipa Rocha

Illustrations: Filipa Rocha



Science4you

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SAFETY RULES

- Read these Instructions before use, follow them and keep them for reference.
- Keep young children, animals and those not wearing eye protection away from the experimental area.
- Always wear eye protection.
- Store this experimental set out of reach of children under 4 years of age.
- Clean all equipment after use.
- Make sure that all containers are fully closed and properly stored after use.
- Ensure that all empty containers and/or non-reusable packaging are disposed of properly.
- Wash hands after carrying out experiments.
- Do not use any equipment which has not been supplied with the set or recommended in the Instructions for use.
- Do not eat or drink in the experimental area.
- Do not replace foodstuffs in original container. Dispose of immediately.
- Do not apply any substances or solutions to the body.
- Warnings: All experiments of this kit should be carried out under the supervision of an adult.

GENERAL FIRST AID INFORMATION

- In case of eye contact: wash out eye with plenty of water, holding eye open if necessary. Seek immediate medical advice.
- If swallowed: wash out mouth with water, drink some fresh water. Do not induce vomiting. Seek immediate medical advice.
- In case of doubt, seek medical advice without delay. Take the chemical and its container with you.
- In case of injury always seek medical advice.

ADVICE FOR SUPERVISING ADULTS

- Read and follow these Instructions, the safety rules and the first aid information, and keep them for reference.
- This experimental set is for use only by children over 4 years.
- Because children's abilities vary so much, even within age groups, supervising adults should exercise discretion as to which experiments are suitable and safe for them. The instructions should enable supervisors to assess any experiment to establish its suitability for a particular child.
- The supervising adult should discuss the warnings and safety information with the child or children before commencing the experiments.
- The area surrounding the experiment should be kept clear of any obstructions and away from the storage of food. It should be well lit and ventilated and close to a water supply. A solid table with a heat-resistant top should be provided.
- This experimental set contains colorings. Colorings can stain. Keep it away from objects and fabrics.
- This experimental set contains seeds. The seeds must be kept away from eyes, nose and mouth. In case the seeds come in contact with eyes or mouth, wash with running tap water. In case of rash or irritation seek medical advice.

PROTECTIVE GOGGLES INSTRUCTIONS

General Applications (EN 166/2802-04)

PROTECTIVE GOGGLES MODELS J0-101, J0-128, MERDOR 101, MERDOR 181 and WVC-5470001: Can protect against basic impacts or low energy, but not against dust or liquid splashes or even against hot metals.

PANORAMICS: Against low or medium energy impacts, dust, liquid drops and cart materials. Not against liquid splashes or electric short-circuits.

FRIGILL VISIONS: Can protect against low, medium and high energy, liquid splashes, cast metals and hot solids. Not against dust and gases.

1 - IDENTIFICATION

- Manufacturer identification: JI, MEDOR, MERDOR, WVC/VINCI, or the acronym.
- Optical class, according with EN 166 (Just for neutral lens).

OPTICAL CLASS	SPHERICAL POWER (D) ± 0.03 (Dm) ²	STIGMATIC POWER (D) - D ₂ (Dm) ²	DIFFERENCE IN PRISMATIC REFRACTIVE POWER (ΔP) (Dm) ²		
			Horizontal		Vertical
			External Base	Internal Base	
1	±0.06	0.06	0.75	0.25	0.25
2	±0.12	0.12	1.00	0.25	0.25
3	±0.12 - 0.25	0.25 ²	1.00	0.25	0.25

Notes: D1 and D2 are the refractive powers of the two main meridians.
² The main meridian axes of the two lenses will be parallel between ± 10°.

- Symbol of mechanical resistance with the letters: W without letters Minimal robustness.

S: Robustness increased (Steel ball from 22mm to 5.1 m/s).

F: Impacts of high speed and low energy (Steel ball from 6mm to 45 m/s).

H: Impacts of medium energy (Steel ball from 6mm to 120 m/s).

A: High energy impacts (Steel ball from 6mm to 190 m/s).

- Marking of the frame

Identification of the manufacturer: JI, MEDOR, MERDOR, WVC/VINCI.

- The number of Standard EN 166, which covers this protector

- A letter indicating the field of use (See table)

- Symbol of resistance to particles of high speed:

- Low energy impact: F

- Medium energy impact: B

- High energy impact: A

Note: If protection against high velocity particles and extreme temperature is required, the eye protector used must be marked with the letter T and immediately after the letter of impact (F, B, A). If the letter of impact is not followed by the letter T, the eyewear should be used against particles at high speed and at room temperature.

2 - USE

It must be ensured that the risk present in the working environment corresponds to the field of use of this protective eyewear, which is deduced from the marks printed on them.

To adapt the protective eyewear to user dimensions, you only need to modify the length on the extension tabs. If the goggles have non-adjustable tabs, you must change the curved position of each tab of the tab by hand: this operation is made easier by gently heating the tab first with warm air from a radiator, hot water, etc.

The helmet-adaptive visors are adjusted by inserting the frame at the edge of the helmet and securing it with the rear elastic closure.

All equipment is manufactured with materials that do not produce allergies, irritations, etc. However, they can produce allergic reactions in especially sensitive people.

They are for personal use and therefore should not be used by several users, even after they have been carefully cleaned.

The prescription glasses should be worn under the conditions for which they were prescribed.

Details of uses:

SYMBOL	DESIGNATION	FIELD OF USE/DESCRIPTION
Without Symbol	Basic use	Mechanical hazards not specified
2	Liquid	Liquid (Drop and Splashes)
4	Thick dust particles	Powder particles larger than 5 microns
5	Gaseous and fine dust particles	Gases, vapors, sprays, smoke and dust particles with a size of less than 5 microns
8	Electric Short Circuit Arc	Electric arc due to short circuit in electrical equipment.
9	Hot melt and solid materials	Splashes of molten metals and penetration of hot solids.

3 - MAINTENANCE

3.1 Care and Cleaning

Full protective goggles should be cleaned at intervals with clean, soapy water. Then dry them carefully with a soft, clean and absorbent cloth. For greater cleaning (disinfection) use a dry cloth impregnated with alcohol. In no case should other types of solvents be used.

3.2 Replacement

With prolonged use, these protective glasses can be scratched and chipped and, in the case of plastic lenses, yellowing. Therefore, it is necessary to do a periodic review of the state of the protective eyewear and replace it if it exhibits these defects, although apparently still allow the vision, and replace the product. In any case, this product must be replaced every 5 years.

3.3 Spare Parts

Spare lenses are available for all face screens. There are also spare lenses for binocular glasses and welder glasses.

3.3 Conservation

Protective goggles should be stored in an airy and cool place, avoiding humidity, dirt and dust. It is recommended to use a protector or a plastic bag to transport the glasses. Disposal considerations: must be handled in accordance with local regulations.

In case of emergency call:

9-1-1 or Poison Control: 1-800-233-1222

In case of poisoning by any of the components used in the experiments of this toy, contact your local poison control center or the nearest hospital. Please consult the following link for more information: <https://www.poison.org/>

LIST OF SUBSTANCES SUPPLIED

Blue Coloring

INGREDIENTS: C142090, SODIUM BENZOATE, POTASSIUM SORBATE

Precautionary Statements:

- P202 Do not handle until all safety precautions have been read and understood.
- P233 Keep container tightly closed.
- P234 Keep only in original container.

Chamomile Seeds

Red Coloring

INGREDIENTS: C14255, SODIUM BENZOATE, POTASSIUM SORBATE

Precautionary Statements:

- P202 Do not handle until all safety precautions have been read and understood.
- P233 Keep container tightly closed.
- P234 Keep only in original container.

Yellow Coloring

INGREDIENTS: C19140, SODIUM BENZOATE, POTASSIUM SORBATE

Precautionary Statements:

- P202 Do not handle until all safety precautions have been read and understood.
- P233 Keep container tightly closed.
- P234 Keep only in original container.

Recommendations for substances and mixtures: Do not ingest. Avoid contact with the eyes and mouth. Use only according to the instructions. Store in tightly closed containers. Keep in a cool, dry place. Protect from moisture, direct sunlight and heat sources.

DISPOSAL OF USED SUBSTANCES

Do not dispose of substances and/or mixtures together with household or other waste. Please recycle packaging materials where local recycling programs exist.



Get to know your kit!



Protective goggles



Large test tubes with lids



Measuring cup



Yellow coloring



Pipette dropper



Blue coloring



Red coloring



Chamomile seeds



Magnifying glass



Tweezers



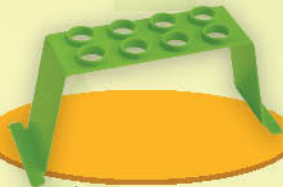
Small test tubes with lids



Flower pots



Bubble wand



Test tube rack



Cotton balls



Card with graphic elements

Scientist, pay attention to what you will need in every experiment!



Note: every time you finish an experiment wash all material and put it away!



Hello, Scientist!
Get ready to explore the world of science! These fun experiments will teach you so many fun and interesting things with your **First Science Kit!**

What is science?

Science means **Knowledge!**



How does science work?

A Scientist has questions!



The Scientist makes experiments!



The Scientist finds answers!



And much more...

What does science study?



Different living beings.



Different musical instruments made of different materials make different sounds!



Planet Earth

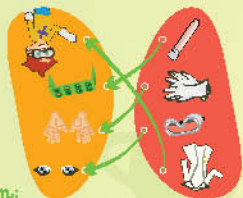
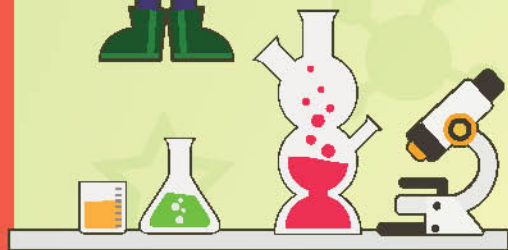
Safety in the laboratory

Game 1.

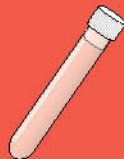
Connect the images on the left to the ones on the right. They will give you some helpful knowledge to always stay safe in the laboratory!



Before we start...
do you know the
main rules of a
scientific
laboratory?



Solution:





Scientist, it is very important that you learn how to properly use the pipette dropper! Are you ready?

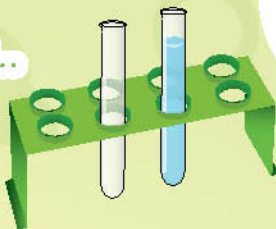
The pipette dropper!



How to do it...

1. Put the 2 test tubes in the test tube rack.

2. Fill 1 of the test tubes with water.



3. Now, hold the pipette dropper, just like you see in the image.



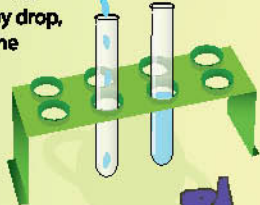
4. Press the upper part of the dropper (rubber top part).



5. Put the pipette dropper inside the test tube with the water and release the rubber top.



6. Put the dropper on top of the empty test tube and let the water fall slowly, drop by drop, while you press the rubber top.



Practice until you are able to drop just small drops! That way you will be an extraordinary Scientist!

Could you do it, Scientist?



What will you need?



Pipette dropper

2 Large test tubes



Test tube rack



Extra materials:

Water

Did the water enter the dropper, Scientist?



It is extremely important that you wash the pipette dropper every time you use it and every time you change reagents (colorings)! To wash it, take out the rubber part and rinse the plastic part several times with water.



Theme: Colors

Experiment 1 Color mixtures



It is from the mixture of these three colors that others are obtained, just like the colors of the rainbow.

Red

Yellow

In nature, there are 3 primary colors.

Blue



To know the primary and secondary colors and learn how to obtain the secondary colors, using just the primary colors.

What will you need?

Pipette dropper

3 Large test tubes with lids

3 Small test tubes with lids

Test tube rack

Measuring cup



Extra materials:

Water

Gloves

1. Put on the gloves.



2. Put the test tubes in the rack, as the image shows.



3. Using the measuring cup, fill the 3 large test tubes with water.



4. Let's add color to the water in each test tube! With the dropper, add 5 drops of a coloring to one of the large test tubes. Repeat the steps for the remaining test tubes, with the rest of the colors.



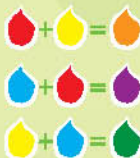
Are you ready to make new colors?



5. Next, carefully put one of each color into the small test tubes. Then, use the color sheet in the image to the right to choose a color combination. Put the second color in the small test tubes and mix them together. Finally, put the lids on the small test tubes to save the colors for later.



What do you observe?
Which colors can you create?



In this experiment, it was possible to observe that, when we mix two colors (primary), we get a new color (secondary).

Did you know...
That if we add secondary colors to a primary color, we obtain a tertiary color? Brown is an example of a tertiary color because it is the mixture of green with red!



Conclusion!

Theme: Living things - plants

Plants are living things that use the Sun's energy and carbon dioxide for food. They are extremely important because they produce the oxygen that we breathe!



Tree



Plant with flower



Non-flowering plant



Learning how plants grow and develop.



Chamomile

Experiment 2 A garden of flowers

Did you know...
That there are several types of plants?

What will you need?



Extra materials:

- Cup • Water • Scissors
- Gloves

Always ask an adult for help!



How to do it...



4. Ask an adult to help you use the scissors to cut out the flower tags on the card with graphic elements. Write the name of the plant on the tag and put it in the flower pot to identify it.

5. Keep watering your flower pot with the dropper. Don't forget that plants also need sunlight to grow.



Now you just need to wait until the plants start to germinate! When they grow, you can plant them in larger flower pots or in the garden.



Conclusion!

Plants germinate (develop) from seeds! To grow, they need nutrients, water, and sunlight. But not in excess or they can become sick and die.



1. Put on the gloves and put 3 cotton balls in the flower pot.

4 to 10x

2. Put between 4 to 10 chamomile seeds in the flower pot.



3. Put a bit of water in a cup and then water the flower pot using the dropper.



You can also use soil for plants instead of cotton balls to perform this experiment.



Chamomile

Ideal period to plant seeds:

Spring

Blooming season:

During spring and summer

Experiment 3

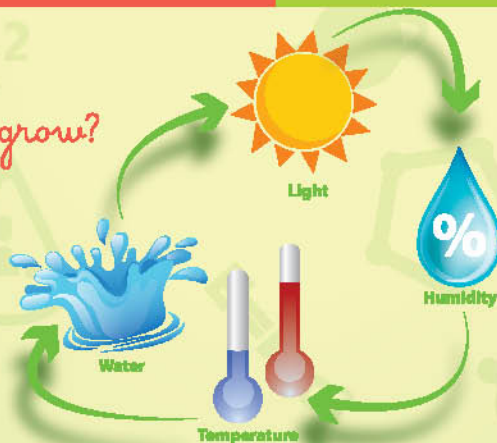
What do plants need to grow?



A plant's life is influenced mainly by four factors:



To know and observe the factors which influence a plant's growth and survival.



What will you need?



Pipette dropper



Cotton balls



Extra materials:

- 3 Cups • Marker • Water
- Grass seeds • Gloves



How to do it...

1. Put on the gloves and use a marker to identify the 3 cups with A, B and C.
2. Put a cotton ball in each cup.
3. With the dropper, put between 5 to 10 drops of water in each cup.
4. Put some grass seeds in each cup.
5. Place the cups as follows:



You just moistened the cotton!



Cup A

Near a window and keep moistening it.



Cup B

In a dark place and keep moistening it.



Cup C

Near a window, without adding more water on this cup.



What happened in the 3 cups, Scientist?



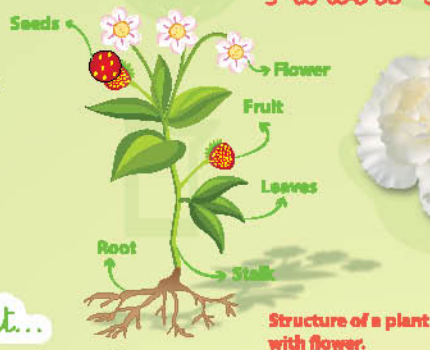
Conclusion!

Grass should only have grown in cup A. We can learn that plants can't grow without light or water!

Experiment 4

Flowers that change color

You already know that there are several types of plants. A plant's structure is interesting. Have a look at the example!



To understand how plants get food.

What will you need?



Extra materials:

Plant with a white flower (e.g. carnation)
Pitcher • Scissors • Spoon • Water • Gloves

Always ask an adult for help!



1. Put on the gloves and put 200 milliliters (ml) of water inside the measuring cup.



2. With the dropper, add 15 drops of your favorite color to the water. Stir with the spoon.



3. Ask an adult to cut the tip of the white flower stalk at an angle.

4. Put the colored water in a pitcher.



The measuring cup has a scale (ruler) which helps you measure the quantity (volume) of the liquids you put inside.

5. Put the flower in the pitcher.



Wait a few days and see what happens. Keep adding water to the pitcher so the flower is never without water.



SUPER SCIENTIST:

Use the colorings you like the most and get flowers of many different colors. Perform experiment 1 and then use the new colors that you created.

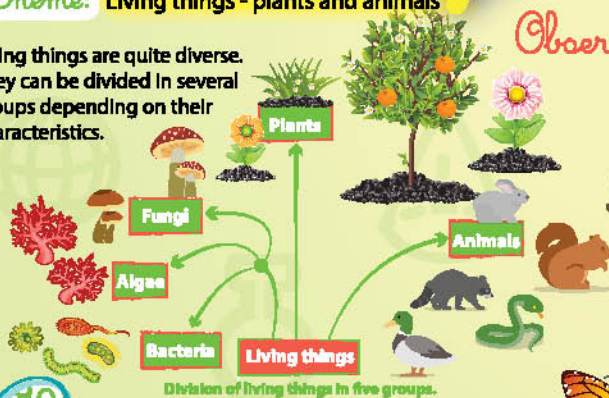


Conclusion!

The colored water goes up the stalk until it reaches the leaves and flower, which turn the color that you added to the water. Plants feed through their roots. In the absence of roots, the stalk transports the water directly to the plants.

Theme: Living things - plants and animals

Living things are quite diverse. They can be divided in several groups depending on their characteristics.

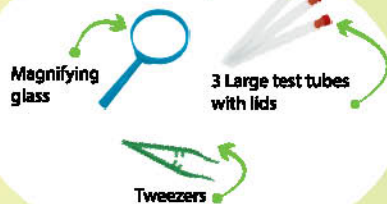


Experiment 5 Observing living things



To observe some of the characteristics of living beings.

What will you need?



Extra materials:

- Flower from experiment 2 or 4
- Plants • Insects • Notebook • Pencil



How to do it...

Observing plants

1. Choose the plants you want to observe.

Look for them in a garden or in your house!

3. With the tweezers, remove some leaves and/or petals and observe them.

4. Observe also the flower of experiment 4.
5. Draw what you observe in your notebook.



Conclusion!

This is an excellent technique to observe living things in their own environment. That way, it is possible to see how they behave and the great diversity of living things that exists!



How to do it...

Observing insects

1. Take your material outside on a sunny day.
2. Look for insects on your plants.
3. When you find an insect try to hold it with the tweezers and put it in the test tube.
4. Cover the test tube with the lid.
5. With the magnifying glass, observe all the insects you wish and draw what you see. *Can you observe 6 legs and 2 antennae?*
6. In the end, put the insects back in nature.



Don't keep insects for longer than 2 minutes inside the closed test tube!

Theme: Art with science

By using certain materials' properties, it is possible to create mixtures which become new materials.

Paints are the result of mixing several components.

There are so many types of colorful paints!

Paints are an example of that!

Experiment 6 Finger paint



To make paint that can be used with the fingers, using only simple materials!

What will you need?



Dropper pipette



Colorings



Measuring cup

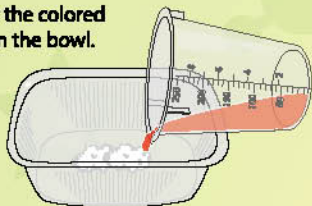


Extra materials:

Flour • Water • Tablespoon
Bowl • Sheets of paper



4. Put the colored water in the bowl.



1. Start by putting 2 spoons of flour in a bowl.



2. Measure 20 ml of water with the measuring cup.



5. Mix well with your fingers.



3. Use the dropper to add some drops of your favorite coloring to the water.



6. Repeat these steps using different colors, or mix the ones you already made.



Now have fun and make fantastic works of art on sheets of paper!

ATTENTION: when you finish the experiment, throw away all used food.

Do you remember experiment 1? What do you observe, Scientist?



Conclusion!

In this experiment you made fantastic paints using just ingredients from your everyday life. This is only possible due to the properties of the materials you used!

Scientist, use these paints in the following experiment!

Theme: Living things - fingerprints

Fingerprints are the drawings which you can see on the tip of your fingers.



These are different for every person!

Experiment 7 Art with fingerprints

Did you know...
That not even twins have the same fingerprints?
That is why they are used to identify people!



What will you need?



Magnifying glass



Paints made in experiment 6



Extra materials:

- Sheets of paper



Goal

To observe fingerprints, and to understand that there aren't two identical in the world.



How to do it...

1. Prepare the sheets of paper.

2. Choose a paint and put your Index finger in it. Then press your finger against the sheet of paper.

Observe fingerprints!

3. Use the magnifying glass to see the details of the drawing that remained on the paper.

What do you observe? This is your fingerprint!

4. Repeat the experiment but now ask other people to press their fingertips to the paper.

Are their prints the same as yours?

Create art with your fingerprints!

Try to create drawings, using just the tip of your fingers (your fingerprints)! Use your imagination!

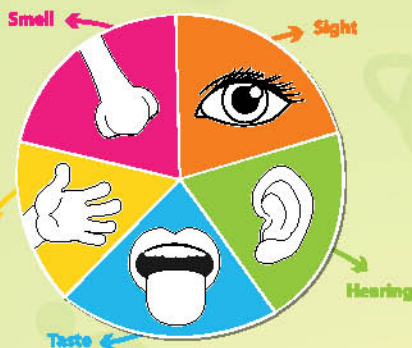
Conclusion!

Fingerprints are very important. Everyone has a different pattern that helps to identify them!

Theme: Living things - The five senses



Human beings have five senses:



1. Put the test tubes in the rack.
2. Ask an adult to put a spice (cinnamon, curry, coffee chocolate, vanilla, etc.) in each tube without letting you see it.
3. Put on the blindfold and then smell each one of the tubes.
4. Wash the test tubes thoroughly with water and soap.
5. Repeat the experiment but now ask an adult to put some drops of liquid fragrances (such as different perfumes or different fruit juices) on the cotton balls.

ATTENTION: when you finish the experiment, throw away all used food.

Experiment 8

What is that smell?

To recognize the smell of your lunch or of a perfume, we use the **sense of smell**!



To recognize different fragrances using the five senses of the human body.



Cinnamon



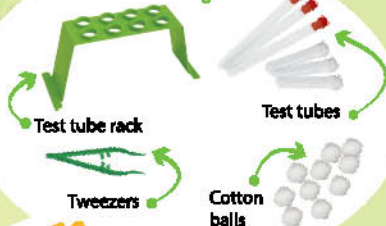
Curry



Vanilla



What will you need?



Extra materials:

- Blindfold • Different spices and perfumes
- Fruit juices • Other things you want to test

Always ask an adult for help!



Conclusion!

The nose is the organ responsible for the sense of smell. Even with the eyes blindfolded, it is possible to distinguish different smells and identify them! That happens because the human body has what we call an olfactory memory.

Experiment 9

The flavor game

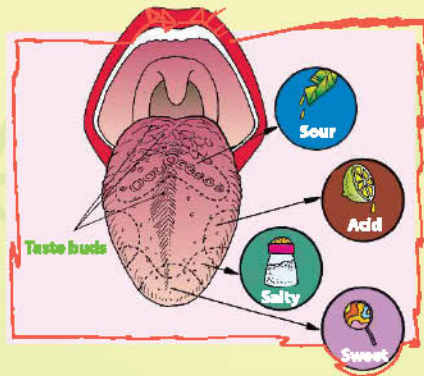
What will you need?

 **Extra materials:**

- Opaque cups • Water • Lemon juice
- Chocolate powder • Coffee powder
- Table salt

Always ask an adult for help!

The sense of **taste** can recognize 4 different main types of flavors.



To recognize the 4 main flavors detected by the human tongue.



2. Ask an adult to put a teaspoon of each of the following ingredients in each cup, without you looking.



3. The adult should then make you taste each cup, one at a time.

ATTENTION: when you finish the experiment, throw away all used food.

1. Fill 4 opaque cups half full with water.

You can also ask an adult to give you other flavors for you to guess.

4. Can you guess the flavors of each cup?



Repeat the experiment but cover your nose. Do you experience all the flavors in the same way?



The human tongue detects 4 different types of flavors! Furthermore, taste and smell act simultaneously, helping us experience all the flavors.



Theme: Water

Water is essential to our everyday life. Do you know the water molecule? It is made of 3 atoms!



Molecules are like building bricks which join together to create all the things you see around you!



Experiment 10 Giant soap bubbles

What will you need?



Extra materials:

- Dish soap • Water • Gloves
 - Liquid glycerine • Shallow dish or plate
- Always ask an adult for help!**



How to do it...



To learn how to make giant soap bubbles!

1. Put on the gloves and goggles.



2. Add 100 ml of water to the measuring cup.



3. Add 50 ml of dish soap to the measuring cup.



4. Add 50 ml of liquid glycerine to the measuring cup.



The liquid will be on the 200 ml mark.

Don't forget to check the measuring cup scale!



You just created a solution for soap bubbles!

5. Stir the solution and then pour it carefully onto the plate.

6. Now, you only need to use the soap bubble hoop to make your giant soap bubbles!



The liquid should now be where it says 150 ml.



Conclusion!

Dish soap and liquid glycerine increase the power between the water molecules, giving them more elasticity. That is why we are able to make giant soap bubbles.

Theme: Water & Art with Science

Water has 3 properties:



It has no smell



It has no color



It has no flavor



How to do it...

1. Put on the gloves and goggles and put a bit of soap bubble solution in a bowl.
4. Prepare a sheet of paper to make your paintings.
6. Blow through the other tip of the straw. Can you make soap bubbles?
7. Blow onto the sheet of paper and see what happens!



Experiment 11 Painting with soap bubbles

When we mix something with water we are making a solution!



To use science to create fantastic works of art with soap bubbles.

2. Choose a color and, with the dropper, add 5 or 6 drops to the bubble solution.



5. Use the straw and dip one of the tips in the colored solution.



8. You can also use more than one straw at the same time.

9. If you want, you can still use the soap bubble hoop.



What will you need?



Extra materials:
• Bowls • Straw • Paper sheets
• Gloves

Repeat these steps and make solutions of the colors you wish.
Do not forget to wash the dropper every time you change colors.



Conclusion!

The colors of the colored solutions stay marked on the sheets of paper!

ATTENTION: always blow through the same side of the straw(s) and when you finish the experiment throw them away.

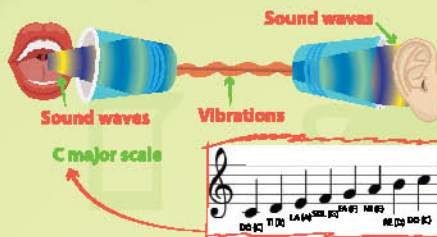
You can save these solutions in the test tubes, covering them with the lids, or in other jars with lid.



Theme: Sound

Sound is the result of a **vibration**: the vibration of particles present in the medium.

These vibrations are captured by the **sense of hearing**, that is, by our ears.

Experiment 12
Water xylophone

To simulate music notes with a xylophone made with water.

What will you need?



Extra materials:

- 8 Glass jars of the same size
- Metal spoon • Water • Gloves



How to do it...

1. Fill two glass jars with the same amount of water. One of them will be our control jar and the other the musical note TI (B).

Attention, you must always use the measure of the control jar to fill all the eight glass jars.



2. Fill a third jar with 2x the amount of water of your control jar.



3. Fill a fourth jar with 3x the amount of water of your control jar.



4. Fill a fifth jar with 4x the amount of water of your control jar.



5. Fill a sixth jar with 5x the amount of water of your control jar.



6. Fill a seventh jar with 6x the amount of water of your control jar and an eighth jar with 7x the amount of water of your control jar.



8. Use the metal spoon handle and gently click the jars with it. Have fun playing and making new sounds!

7. As mentioned in step 1, the jar containing only 1 x the amount of water of your control jar is the musical note TI and the one containing 7x that measure is the Do (C).



Remember experiment 1?
With the pipette dropper, use
the colorings and add different
colors to each music
note (each jar).

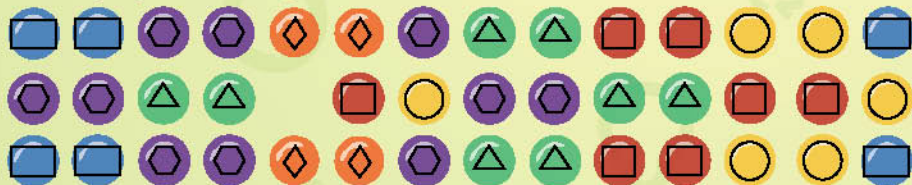


Do (C) Re (D) Mi (E) Fa (F) Sol (G) La (A) Ti (B)

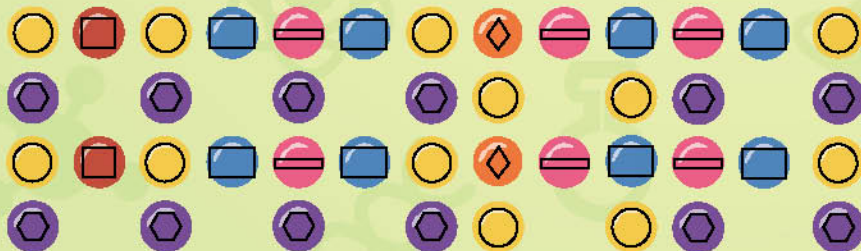


Different quantities of
water create different
sounds due to the
vibration of molecules.

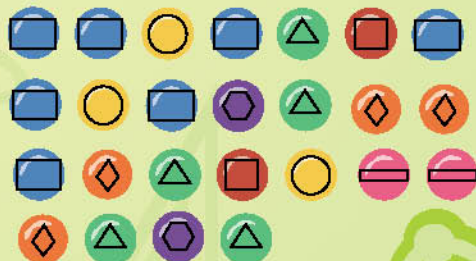
Play fun songs! Twinkle, twinkle little star



London bridge is falling down



Happy birthday song



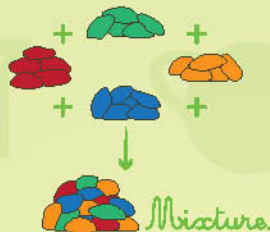
Experiment 13

Theme: Mixtures

When we put together two substances,
we say we are making a *mixture*.

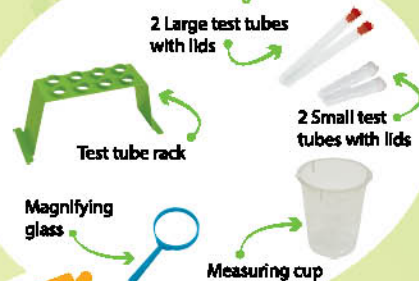
Chemistry is the science that tells us of what
things are made and how they interact together.

(Im)possible mixtures



To understand that not
all substances can be
mixed together and to
observe different types
of mixtures.

What will you need?



Extra materials:

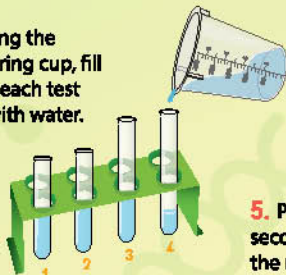
- Water • Olive oil • Sand • Lemon juice
- Powdered chocolate • Teaspoon • Gloves



1. Put the test tubes
in the test tube rack.

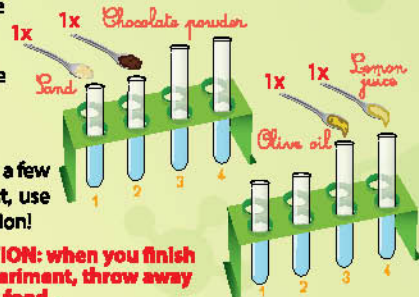
2. Using the
measuring cup, fill
half of each test
tube with water.

4. Cover
the test
tubes with
its lids and
stir them,
one by one.



3. Put on the gloves and add the
solid components to the small
test tubes and the liquids to the
large test tubes.

5. Put the test tubes back in the rack. Wait a few
seconds and observe the results. Scientist, use
the magnifying glass for a better observation!



ATTENTION: when you finish
the experiment, throw away
all used food.

Do all the
substances
mix with water?



In the test tubes 1 and
3, the substances **do not** mix. In the test
tubes 2 and 4, the
substances **mix**.



Conclusion!

Not all substances mix with water. We can say that
there are two types of mixtures in nature: **homogeneous**
mixtures — when substances mix — and **heterogeneous**
mixtures — when substances do not mix.

Theme: Density

Substances' properties make them behave in a special way.

Density is like the "weight" of substances. The ones that float are less dense than the ones that are on the bottom!



Iceberg

Experiment 14 Colored Icebergs

Do you know what an iceberg is?
An iceberg is a great mass of ice which floats in water.
How is that possible?

It is possible thanks to its **density!**



What will you need?



Colorings



Measuring cup



Pipette dropper



Extra materials:

- Ice tray • Water • Cooking oil • Spoon
- Tall transparent cup • Gloves

Always ask an adult for help!



Goal
To observe the density of the substances and how it influences their behavior.



How to do it...

1. Put on the gloves and fill the measuring cup with water.



2. Choose the coloring that you want to use and, with a dropper, add between 3 and 4 drops to the water. Stir well with a spoon.



5. After this time, fill $\frac{1}{2}$ of the tall transparent cup with water.



6. Next, add the cooking oil to the cup.



Ice
Oil
Water



Higher
Density

Water in the liquid state is denser than in the solid state (ice).



3. Ask an adult for help and pour the colored water in the ice tray. Then, the adult should put the ice tray in the freezer.



4. Wait at least 1 hour until the ice cubes form.



Conclusion!

In this experiment it is possible to observe that water and oil don't mix, and that the ice cube floats in oil. As the ice cube begins to melt, the colored water drops start descending to the bottom of the cup. That is due to different densities.

Different objects have different weights. Depending on the weight and the force they exert on the water, they can float or sink.

That always depends, as you already know, on density!

Experiment 15

Does it sink or float?



Extra materials:

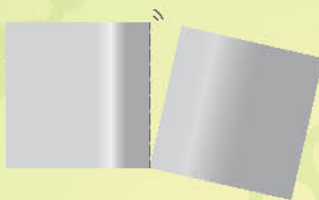
- Aluminium foil
- Large bowl
- Water
- Marble
- Cork
- Coins
- Gloves
- Other materials you would like to test



To understand that some objects are able to float.



1. Tear the aluminium foil to make two square pieces of the same size. Ask an adult to help you.



2. Make a small ball with one of the squares.



3. Fill the large bowl with water.



Are you ready to test what floats and what sinks?

4. Put all of your test materials in the bowl to see what happens!



Conclusion!

Denser objects sink. That happens because water can't make the necessary force to keep the objects floating.

Can you identify which materials float and which sink?



Theme: Solutions and densities

Density is the result between a body's mass and volume!

A full soccer ball (higher volume) floats in water but a deflated soccer ball (lower volume) sinks.

Why?

That happens because, as its mass is equal, its density increases when the volume is lower.

How to do it...

1. Put on the gloves and goggles.

2. Use the teaspoon and put sugar in the cups, just like you see in the image:



4. Mix well with the spoon to dissolve the sugar (solute).

5. With the dropper add a few drops of one color to each cup.



Do you remember experiment 1, Scientist?



Can you create a colored cup?



Experiment 16 Sugar rainbow

Did you know...
That the substances dissolved (mixed) in liquids are called **solutes**?

To observe that the same quantity of water can have different densities if the quantity of the solute (mass) is different.



3. Ask an adult to heat water, and with the measuring cup, add 50 ml of water to each cup.



6. Ask an adult to help you pour in the cup which has 8 spoons of sugar (cup 4) the rest of the solutions, following the order you see in the image.

ATTENTION: when you finish the experiment, throw away all used food.



Conclusion!

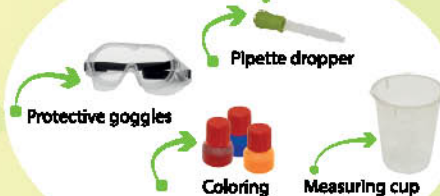
The more the sugar there is in the same quantity of water, the higher the density of the solution (as more mass exists in the same volume).

Less dense

Denser



What will you need?



Extra materials:

• Sugar • 4 Cups • Teaspoon • Gloves
Always ask an adult for help!

Experiment 17

The dance of colors

Theme: Surface tension

Molecules, as you know, bind to each other, creating forces.

These forces are responsible for allowing the materials to present their characteristics.



On the surface of liquids, molecules create a force called *surface tension!*



How to do it...

4. "Prick" the drops of coloring with the toothpick. Do it to one drop at a time. Try not to move the milk!



1. Put on the gloves and goggles and add a bit of milk to the deep plate.



Scientist, can you make the colors dance?



Did you know...

That it is due to surface tension that water scorpions and other insects are able to walk on water?



To observe the force exerted in the surface of liquids.

3. Dip the toothpick in a bit of dish soap.



2. With the dropper, add some drops of coloring to the milk.

What will you need?



Pipette dropper



Protective goggles



Coloring



Extra materials:

• Toothpick • Milk • Deep plate
• Dish soap • Gloves

Always ask an adult for help!



Conclusion!

Milk is, just like the rest of the materials that exist, made of **molecules**. These molecules bind in all directions. However, on the surface of the milk (liquid) these forces form a kind of barrier. Dish soap breaks this barrier, creating the dance of colors you observe!

ATTENTION: when you finish the experiment, throw away all used food.

Theme: Chemical reactions

Have you noticed how some coins are shiny and others are dark?

That happens because, with time, chemical reactions occur, which **oxidize** coins (that is, the material from which these are made — copper).



Copper

Experiment 18

Shiny coins



To learn how to use chemistry to polish coins.

What will you need?



Protective goggles



Magnifying glass



Tweezers



Extra materials:

- Paintbrush • Vinegar • Ketchup
- 2 Plates • Rusty copper coins • Gloves

How to do it...

1. Put on the gloves and goggles.

2. Use the magnifying glass to thoroughly observe your rusty copper coins.

3. On a plate, put a bit of vinegar and in another a bit of ketchup.

5. With a paintbrush, rub some of the ketchup and vinegar on the coins.

6. Take the coins off the plates with the tweezers and then observe them!

4. Use the tweezers to put a coin on each plate.

ATTENTION: when you finish the experiment, throw away all used food.

What do you think, Scientist?
Are your coins shinier?



Conclusion!

Vinegar and ketchup have acid compounds in their composition. These compounds react with oxidized copper, cleaning the coins.

If you try to clean your coins with water, for example, you will not get the same result because water is not acid!

Experiment 19
Hidden colors

Theme: Mixture separation

As you know, colors can be the result of a mixture between other colors. Would it be possible to separate the colors again?

In laboratories, scientists use many techniques to separate mixtures. One of them is

chromatography!

To use the chromatography technique to separate secondary colors from the primary colors that created them in the first place.

What will you need?



Extra materials:

- Paper filter (coffee filter)
- Water
- Colored markers
- Napkin



How to do it...

1. Make several dotted circles on the paper filter with the colored markers.

3. When the paper filter is colorful, it is time to separate the colors.

4. Twist the tip of the napkin and dip it in the cup of water.



2. Make a circle of dots using mainly secondary or tertiary colors.

Secondary colors

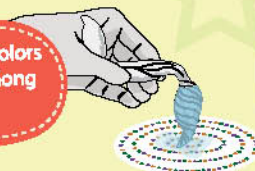


Tertiary colors



5. Finally, put the wet tip of the napkin in the center of the paper filter and let it absorb the water.

Can you see the colors being dragged along the paper?



Do you still remember which are the secondary colors?



Conclusion!

Chromatography separates the colors along the paper filter because it is porous. That way, colors are separated depending on the weight of the molecules.

Theme: Crystals

There are 2 types of solids in nature: **crystalline** and **amorphous**.

The difference between them is the way their molecules organize themselves.



Experiment 20

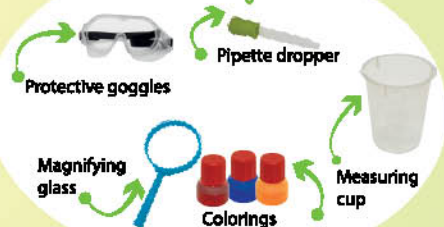
Crystalline messages



To learn how to create a saturated solution which, with time, creates crystals.



What will you need?



Extra materials:

- Sea salt • Spoon • Hot water • Gloves
- Cup • Black cardboard • Paintbrush

Always ask an adult for help!

How to do it...

1. Put on your protective goggles and gloves. Measure 50 ml of sea salt with the measuring cup and put it in the other cup.



2. Use the measuring cup again to measure 100 ml of hot water.



4. Finally, add the salt and mix with a spoon.



You just created a saturated solution of salt!

5. Use the paintbrush and draw whatever you want on the black cardboard. You can write words or make drawings.

3. Use the dropper and add 3 drops of a coloring to the water.



Note: It is important that you keep mixing the solution and that you wet the paintbrush many times!

Now you only need to wait until the water evaporates and beautiful crystals of salt are formed on the paper. The color of your crystals will be determined by the coloring you choose. You can make solutions with many different colors, Scientist!



Conclusion!

Initially, the salt dissolved in the water. As the water evaporates, the molecules of the dissolved salt join again in an organized structure, creating beautiful salt crystals!



Experiment 21

Home-made modeling clay

Theme: Moldable materials

In nature, not all materials behave the same way. That will depend on the way their molecules bind.

Clay or dough is a moldable material.

Glass is a hard material.

Stronger bonds create harder materials and vice versa.



To create home-made modeling clay, using everyday ingredients.

What will you need?



Extra materials:

- Flour • Cooking oil
- Water • Spoon • Bowls



How to do it...

4. Finally, add 20 ml of cooking oil to the previous mixture (use the measuring cup).

1. Start by putting 10 full spoons of flour in a bowl.

10x



5. Mix all thoroughly until you get a homogeneous mixture. The home-made modeling clay is now ready!

2. In a measuring cup put 100 ml of water. Then, with the dropper, add some drops of a coloring of your choice.



3. Add the colored water to the flour.



Repeat the experiment using other colorings and make modeling clay of any color!



Conclusion!

When the flour properties mix with water, they create this fantastic modeling clay! That happens because wheat forms a dough called gluten!



Wheat



Theme: Viscosity - Non-Newtonian fluids

Viscosity is a property of fluids (liquids and gases). The more the viscosity, the more it is difficult for the fluid to move.

Some fluids, however, behave in an opposite way. These are called

non-Newtonian fluids.

Ketchup is a non-Newtonian viscous fluid.



Experiment 22 Crazy dough



To observe how corn starch behaves when it is mixed with water.

What will you need?



Pipette dropper



Colorings



Measuring cup



Extra materials:

- Corn starch
- Bowl
- Gloves
- Tablespoon
- Water



How to do it...

1. Put on the gloves and add 5 full tablespoons of corn starch to the bowl.

5x



2. In the measuring cup put 50 ml of water.



50 ml

ATTENTION: when you finish the experiment, throw away all used food.

3. Choose a color, and with the dropper, add some drops to the water.



4. Add the water to the corn starch and mix well with your hands.



Shake the dough slowly and observe that it behaves as if it were a liquid.

Try to make a small ball. Can you do it? What happens when you release it?

Poke the surface of the dough and see how it appears to be solid!

Notes:

- If it gets too dry, add more water
- If it gets too liquid, add more corn starch

Your crazy dough is ready. Test it!



Conclusion!

This crazy dough is a non-Newtonian fluid. That means it has the opposite behavior of normal fluids: the more force is applied to it, the harder it gets.

Experiment 23

Home-made sticky monster

Theme: Polymers

When many molecules bind in a very long chain, we say we have a

polymer.

Plastic materials, for example, are made of

polymers.



Monomer



Polymer



To create a fun and sticky dough, using everyday materials.

What will you need?



Measuring cup



Extra materials:

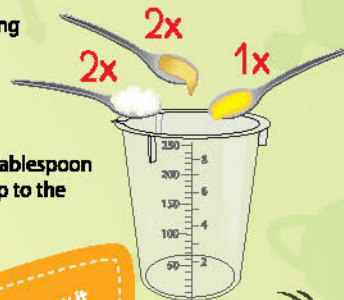
- Corn starch • Dish soap • Gloves
- Cooking oil • Tablespoon

Always ask an adult for help!



How to do it...

1. Put on the gloves and then add 2 tablespoons of corn starch to the measuring cup.
2. Add 2 tablespoons of cooking oil.
3. Stir well until the mixture becomes homogeneous.
4. Add a tablespoon of dish soap to the mixture.



Make small balls with the dough or mold it as you wish.

Take the dough you made from the cup and have fun with your homemade sticky monster!

ATTENTION: when you finish the experiment, throw away all used food.

Conclusion!

The dish soap made this dough sticky, allowing you to stick it to the wall.

You can also throw it against a wall. Does it stick there, Scientist?

Theme: Light

A rainbow results from the separation of the sunlight (white light).



Red Orange Yellow
Green Blue Indigo Violet

Experiment 24

The colors of the rainbow



Observe what happens when the 7 colors of the rainbow are mixed.

What will you need?



Newton disc (card with graphic elements)



Extra materials:

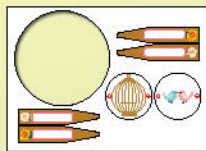
Scissors • Pencil • Adhesive tape
Always ask an adult for help!



How to do it...

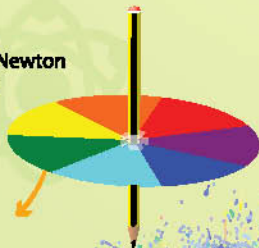


1. Ask an adult to help cut out the Newton disc from the card with graphic elements.

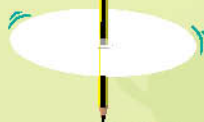


4. Attach the pencil to the Newton disc with a bit of sticky tape.

Your Newton disc is ready!

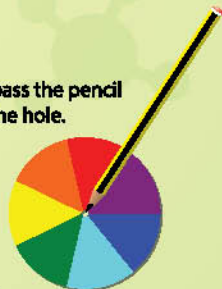


5. Now you need to make the disc spin as if it were a spinning top.



2. Ask an adult to make a hole in the center of the disc using a pencil.

3. Now pass the pencil through the hole.



Conclusion!

When the Newton's disc is spinning like a top, we observe the white color because that is the result of the sum of the colors of the rainbow.

When the disc spins too fast, the eyes and the brain, as they work together, can't distinguish each color separately. That makes it seem that the disc is white.

Did you know...

That rainbows only appear when sunlight is decomposed by the presence of water drops in the air?

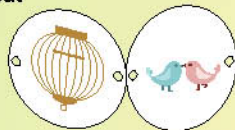
Theme: Optical illusions

It is because of light that our eyes can capture the images that we see.

The sense of **sight** works when light gets inside our eyes, stimulates the nerves and sends signals to the brain, which deciphers the "message."



3. Use the glue stick and glue both circles. Make sure you glue the circles so the pictures are both facing out, and that the pictures are flipped away from each other like the picture here.



5. Twist the card circles to wind the strings.



1. Ask an adult to help cut out the birds and birdcage circles from the card with graphic elements.



4. Ask an adult to help you cut 2 strings about 16 inches each. Then tie 1 string to each side of the card circles.



6. Stretch the strings swiftly and see what happens!

Did you know...
That this "technique" is used in cartoons?

Experiment 25 Cartoons



To observe how both images can look like a single one through an optical illusion.

What will you need?



Circles — birds and birdcage (card with graphic elements)

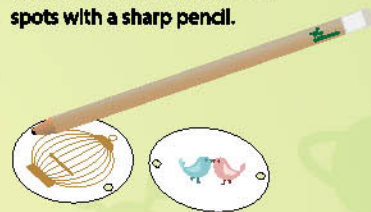


Extra materials:

- Scissors
- Sharp pencil
- Yarn or string
- Glue stick

Always ask an adult for help!

2. Ask an adult to make two holes in each circle in the indicated spots with a sharp pencil.



Conclusion!

When you stretch the strings, the card circles start spinning really fast and it looks like the birds enter the birdcage — an optical illusion occurs.

Theme: Planet Earth



Sand results from rock erosion through the action of wind, water and living things.



Cliffs suffer erosion and create sand.

Experiment 26 Sand painting



To use science to create fantastic works of art!

What will you need?



Colorings



Pipette dropper



Extra materials:

- Beach sand
- Bowls
- Gloves
- Cardboard
- Markers
- Cotton swabs
- Liquid glue
- Spoon



How to do it...

1st part - Prepare the sand

1. Put on the gloves and put a bit of sand in a bowl.



2. Choose a coloring and with a dropper, add 3 to 4 drops to the sand.



3. Mix well with a spoon.

Do you like the color?
Then wait a bit until the sand dries.
If you want, add more coloring (so the color gets stronger) or more sand (so the color gets lighter).

4. Repeat this procedure to create sand of many different colors.



2nd part - Create works of art

1. With the markers, make a drawing on a piece of cardboard (any format).



3. Without letting the glue dry, sprinkle the colored sand you prepared over the drawing.



2. Use a cotton swab to rub the liquid glue over your drawing.



4. Let it dry well and then shake off any extra sand.

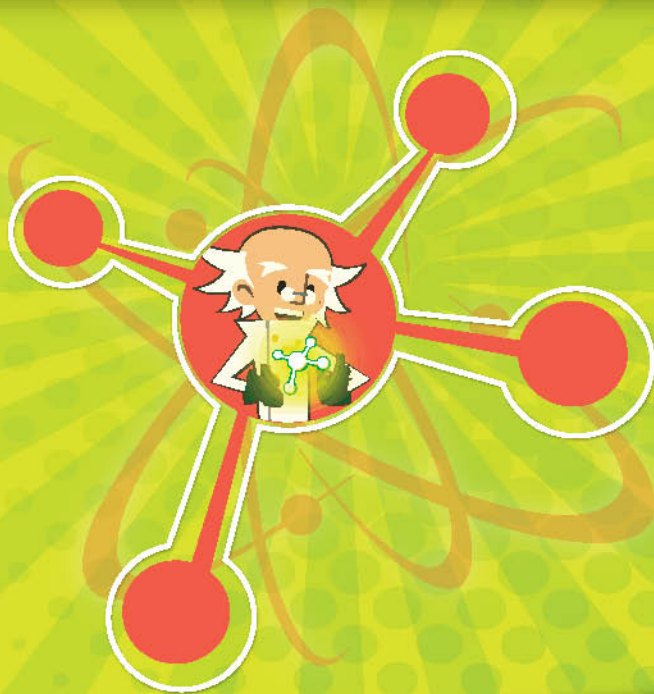


How were your sand paintings, Scientist?



Conclusion!

Nature gives us many materials which we can use to create fantastic works of art. You only have to use your imagination!



For more information visit our website:
www.playmonster.com

